

Magnetic properties of $\text{Cu}(\text{en})(\text{sal})\text{Cl}$: a quasi-two-dimensional $S = 1/2$ Heisenberg ferromagnet on an anisotropic square lattice

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A combined theoretical and experimental study of the magnetic properties of the metal-organic compound $\text{Cu}(\text{en})(\text{sal})\text{Cl}$ (en = ethylenediamine = $\text{C}_2\text{H}_8\text{N}_2$; sal = 2-hydroxybenzoic acid = $\text{C}_7\text{H}_6\text{O}_3$) is presented. The compound realizes a quasi-two-dimensional $S = 1/2$ Heisenberg spin system on an anisotropic square lattice. The crystal structure consists of rectangular-like layered frameworks of the Cu^{2+} ions in the bc plane and features two distinct coordination environments: axially elongated CuN_4Cl_2 octahedra formed by en ligands, and compressed CuO_4Cl_2 octahedra arising from asymmetrically bound sal anions [1].

First-principles *ab initio* calculations predict dominant ferromagnetic nearest-neighbor exchange interactions within the layers along c axis, with an exchange constant $J_1/k_B = 5.69$ K, accompanied by weaker antiferromagnetic coupling $J_2/k_B = -0.24$ K along b axis. To verify these predictions, comprehensive experimental studies of the heat capacity, magnetization, and susceptibility were performed over the temperature range 0.38–300 K and in magnetic fields up to 9 T.

Zero-field specific-heat measurements revealed λ -like anomaly at $T_C = 0.82$ K, indicating a phase transition to long-range magnetic order. This transition was further evidenced by the splitting of zero-field-cooling and field-cooling magnetization curves, as well as by the observation of a ferromagnetic hysteresis loop at $T = 0.5$ K. Comparison of experimental heat capacity and susceptibility data with quantum Monte Carlo simulations of the proposed spin model demonstrates reasonable agreement, establishing $\text{Cu}(\text{en})(\text{sal})\text{Cl}$ as a realization of a quasi-two-dimensional $S = 1/2$ Heisenberg ferromagnet on the anisotropic square lattice.

References:

[1] S. S. Batool *et al.*, J. Struct. Chem. 57 (2016), 1176–1181.

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