

The electronic and magnetic properties of $\text{Nd}_3\text{Co}_{13}\text{B}_2$, $\text{Nd}_5\text{Co}_{19}\text{B}_6$ and $\text{Nd}_5\text{Co}_{21}\text{B}_4$ compounds

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The $\text{Nd}_3\text{Co}_{13}\text{B}_2$, $\text{Nd}_5\text{Co}_{19}\text{B}_6$ and $\text{Nd}_5\text{Co}_{21}\text{B}_4$ compounds were manufactured as promising systems suitable for fabrication of permanent magnets. They belong to the $\text{R}_{m+n}\text{Co}_{5m+3n}\text{B}_{2n}$ family with $(m = 2n, n = 1)$, $(m = 2, n = 3)$, and $(m = 3, n = 2)$, respectively [1-3]. The band structure calculations are performed by the tight binding version of the linear muffin-tin orbital method in the atomic sphere approximation (TB-LMTO ASA) [4-5]. The calculated magnetic moments on Co atoms depend on their local environment and vary in the range of 0.2 to $1.67 \mu_B/\text{atom}$. The calculated values of the total magnetic moments are equal to: 24.02 (20.8 [1]), 29.21 (21.5 [2]) and 31.69 (31.1 [3]) $\mu_B/\text{f.u.}$ for $\text{Nd}_3\text{Co}_{13}\text{B}_2$, $\text{Nd}_5\text{Co}_{19}\text{B}_6$ and $\text{Nd}_5\text{Co}_{21}\text{B}_4$ compounds, respectively, where the experimental values are given in the parenthesis. The main contributions to the total densities of electronic states at the Fermi level are provided by $3d$ electrons of the Co atoms.

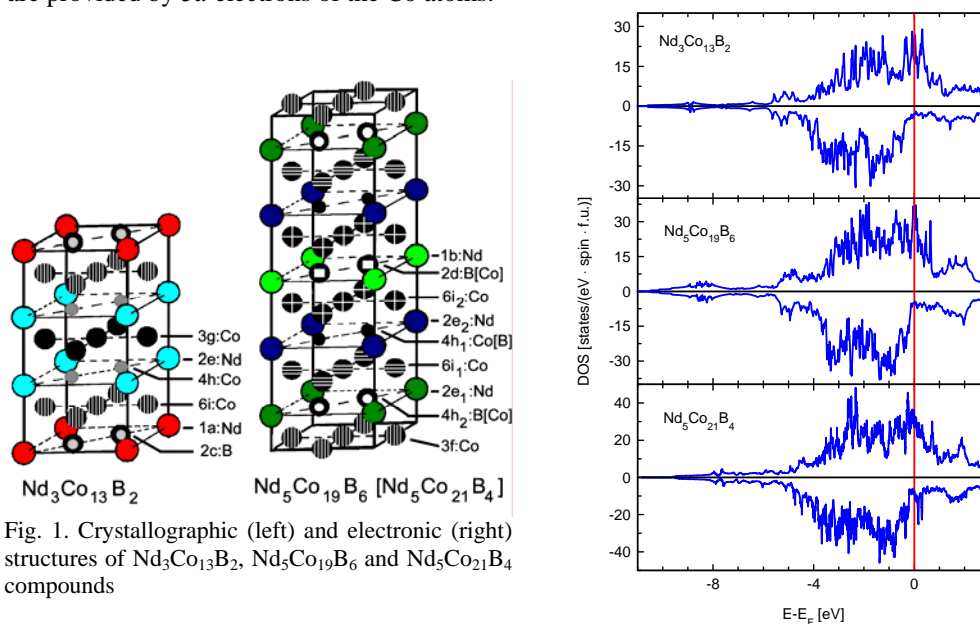


Fig. 1. Crystallographic (left) and electronic (right) structures of $\text{Nd}_3\text{Co}_{13}\text{B}_2$, $\text{Nd}_5\text{Co}_{19}\text{B}_6$ and $\text{Nd}_5\text{Co}_{21}\text{B}_4$ compounds

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